

STIC Search Report

STIC Database Tracking Number, 96758

TO: Michael Feely Location: CP3 6B15

Art Unit : 1712 June 18, 2003

Case Serial Number: 09/899031

From: Kathleen Fuller Location: EIC 1700

CP3/4 3D62

Phone: 308-4290

Kathleen.Fuller@uspto.gov

Search Notes

The structure claims were very broad. Therefore I had to subject search the Chemical Abstracts File, extract the registry number from the pertinent abstracts and search the very broad structure query against the extracted registry numbers.



SEARCH REQUEST FORM

Scientific and Technical Information Center

Mail Box and Bldg/Room Location: 5935 > 6315 If more than one search is submi ***********************************	tted, please prioritiz ********** earch topic, and describe a ywords, synonyms, acron hat may have a special me neet, pertinent claims, and	e searches in order of ****************** as specifically as possible the s yms, and registry numbers, and aning. Give examples or relev abstract.	PAPER DISK E-MAIL need. **********************************
Title of Invention: Liquid Crysle	Comp. Compris	ing Liquid Crystel Mol	scales + Alignment Moment
Inventors (please provide full names):	chibashi, Mits	uyoshiz Kawata, l	(m)
Takeuchi, Hiroshi ; 1			
Earliest Priority Filing Date: 7/	6/2000 (Foren	17/6/2001 (NS)	A STATE OF THE STA
For Sequence Searches Only Please include appropriate serial number.		•	· · · · · · · · · · · · · · · · · · ·
Looking for the all	grand pro	moter in cla	ims
1-16 and	23		
1 10 000			
Thanks,			
			î.
		•	
		•	
·			
		7	· 24
		/	
**************	*******	******	********
STAFF USE ONLY	Type of Search NA Sequence (#)	THE TYPE OF STREET	where applicable
Searcher: K / / / / / / / / / / / / / / / / / /		STN	
Searcher Phone #:	AA Sequence (#)	Dialog	
Searcher Location:	Structure (#)	Questel/Orbit	
Date Searcher Picked Up:	Bibliographic	Dr.Link	
Date Completed: 6/8/03	Litigation	Lexis/Nexis	
Searcher Prep & Review Time:	Fulltext	Sequence Systems	
Clerical Prep Time:	Patent Family	WWW/Internet	
Online Time:	Other	Other (specify)	
PTO-1590 (1-2000)			

EIC1700

Search Results Feedback Form (Optional)



The search results generated for your recent request are attached. If you have any questions or comments (compliments or complaints) about the scope or the results of the search, please contact the EIC searcher who conducted the search or contact:

Kathleen Fuller, Team Leader, 308-4290, CP3/4 3D62

Voluntary Results Feedback Form		
> I am an examiner in Workgroup: Example: 1713		
> Relevant prior art found, search results used as follows:		
102 rejection		
103 rejection	· · · · · · · · · · · · · · · · · · ·	
Cited as being of interest.	• . •	
Helped examiner better understand the invention.		
Helped examiner better understand the state of the art in their t	echnology.	
Types of relevant prior art found:	÷* ∴ ∴ ₹	
Foreign Patent(s)		
Non-Patent Literature (journal articles, conference proceedings, new product announce	ements etc.)	
> Relevant prior art not found:	(* * *	-
Results verified the lack of relevant prior art (helped determine	e patentability).	
Search results were not useful in determining patentability or the	understanding the inver	ntión:
Other Comments:	* * **:	
	•	

Drop off completed forms in CP3/4 - 3D62.

=> FILE REG

FILE 'REGISTRY' ENTERED AT 12:36:17 ON 18 JUN 2003 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2003 American Chemical Society (ACS)

Property values tagged with IC are from the ZIC/VINITI data file provided by InfoChem.

STRUCTURE FILE UPDATES: 17 JUN 2003 HIGHEST RN 532924-24-6 DICTIONARY FILE UPDATES: 17 JUN 2003 HIGHEST RN 532924-24-6

TSCA INFORMATION NOW CURRENT THROUGH JANUARY 6, 2003

Please note that search-term pricing does apply when conducting SmartSELECT searches.

Crossover limits have been increased. See HELP CROSSOVER for details.

Experimental and calculated property data are now available. See HELP PROPERTIES for more information. See STNote 27, Searching Properties in the CAS Registry File, for complete details: http://www.cas.org/ONLINE/STN/STNOTES/stnotes27.pdf

=> FILE HCAPLUS

FILE 'HCAPLUS' ENTERED AT 12:36:22 ON 18 JUN 2003 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2003 AMERICAN CHEMICAL SOCIETY (ACS)

Copyright of the articles to which records in this database refer is held by the publishers listed in the PUBLISHER (PB) field (available for records published or updated in Chemical Abstracts after December 26, 1996), unless otherwise indicated in the original publications. The CA Lexicon is the copyrighted intellectual property of the the American Chemical Society and is provided to assist you in searching databases on STN. Any dissemination, distribution, copying, or storing of this information, without the prior written consent of CAS, is strictly prohibited.

FILE COVERS 1907 - 18 Jun 2003 VOL 138 ISS 25 FILE LAST UPDATED: 17 Jun 2003 (20030617/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> D QUE L26

STR

G2√G1√Cy Si√O Si√O√Ak 4 @5

6 7 @8

broad great the

REP G1 = (0-10) A VAR G2=8/5/CB/AK NODE ATTRIBUTES: DEFAULT MLEVEL IS ATOM

KATHLEEN FULLER EIC 1700/PARKER LAW 308-4290

DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 8

```
STEREO ATTRIBUTES: NONE
L10
            175 SEA FILE=HCAPLUS ABB=ON ALIGN? (3A) PROMOT?
L11
          26249 SEA FILE=HCAPLUS ABB=ON MOLECULAR? (3A) ORIENT?
L15
         140328 SEA FILE=HCAPLUS ABB=ON LIQ?(3A)CRYST?
L16
           8711 SEA FILE=HCAPLUS ABB=ON L15 AND ALIGN?
L17
           3840 SEA FILE=HCAPLUS ABB=ON L11 AND L15
L20
          11523 SEA FILE=HCAPLUS ABB=ON L16 OR L17
L21
                SEL L20 1-11523 RN:
                                        21049 TERMS
L22
          21045 SEA FILE=REGISTRY ABB=ON L21
L24
          15903 SEA FILE=REGISTRY SUB=L22 SSS FUL L4
L25
        1612865 SEA FILE=HCAPLUS ABB=ON L24
L26
             18 SEA FILE=HCAPLUS ABB=ON L10 AND L25
```

=> D L26 BIB ABS HITIND HITSTR 1-18

L26 ANSWER 1 OF 18 HCAPLUS COPYRIGHT 2003 ACS

AN 2003:28027 HCAPLUS

DN 138:391305

TI Colloidal behaviour in a confined cylindrical cell

AU Bowen, W. Richard; Wilson, Jonathan

CS Centre for Complex Fluids Processing, Department of Chemical and Biological Process Engineering, University of Wales, Swansea, SA28PP, UK SO Colloids and Surfaces, A: Physicochemical and Engineering Aspects (2003),

213(1), 59-68 CODEN: CPEAEH; ISSN: 0927-7757

PB Elsevier Science B.V.

DT Journal

LA English

The interactions of colloidal particles in a confined cylindrical cell AB have been investigated using a combination of video microscopy, real-time digital imaging and still micrographs. Confinement of colloidal latex particles of diam. 4.5 .mu.m in cylindrical glass capillaries of diam. 25 .mu.m resulted in their alignment along the center of the cell with a clear particle-free zone along the wall. This tendency was obsd. over a range of ionic strengths from deionized water to 0.1 M NaCl, but was greatest at low concns. of electrolyte. The dimensions of the particle-free zone were orders of magnitude greater than the Debye lengths of the solns. Particles of diam. 3.0 .mu.m also exhibited alignment, but not 1.2 .mu.m particles, indicating that there is a fine balance between the forces promoting alignment and kinetic effects. The particles along the center of the capillary formed dynamic chains in which an av. of .apprx.3 particles were in contact at any time. The rate of movement of the particles increased with decreasing salt content. These observations are consistent with asymmetries in electrostatic or hydrodynamic interactions induced by the cell wall, or possibly by a combination of such effects.

CC 66-4 (Surface Chemistry and Colloids)

IT 9003-53-6, Polystyrene

RL: PEP (Physical, engineering or chemical process); PYP (Physical process); PROC (Process)

(colloidal behavior in a confined cylindrical cell)

CRN 100-42-5

CMF C8 H8

 $H_2C = CH - Ph$

RE.CNT 19 THERE ARE 19 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L26 ANSWER 2 OF 18 HCAPLUS COPYRIGHT 2003 ACS

AN 2002:884235 HCAPLUS

DN 138:98481

TI Effects of Surface Morphology on the Anchoring and Electrooptical Dynamics of Confined Nanoscale Liquid Crystalline Films

AU Noble, Alison R.; Kwon, Hye J.; Nuzzo, Ralph G.

CS Department of Chemistry and the Frederick Seitz Materials Research Laboratory, University of Illinois, Urbana, IL, 61801, USA

SO Journal of the American Chemical Society (2002), 124(50), 15020-15029 CODEN: JACSAT; ISSN: 0002-7863

PB American Chemical Society

DT Journal

LA English

AB The orientation and dynamics of two 40-nm thick films of 4-n-pentyl-4'-cyanobiphenyl (5CB), a nematic liq. crystal, were studied using step-scan FTIR spectroscopy. The films are confined in nanocavities bounded by an interdigitated electrode array (IDA) patterned on a Zn selenide (ZnSe) substrate. The effects of the ZnSe surface morphol. (specifically, two variations of nanometer-scale corrugations obtained by mech. polishing) on the initial ordering and reorientation dynamics of the elec.-field-induced Freedericksz transition are presented here. The interaction of the 5CB with ZnSe surfaces bearing a spicular corrugation induces a homeotropic (surface normal) alignment of the film confined in the cavity. Alternately, when ZnSe is polished to generate fine grooves along the surface, a planar alignment is promoted in the liq. cryst. film. Time-resolved FTIR studies that enable the direct measurement of the rate consts. for the elec.-field-induced orientation and thermal relaxation reveal that the dynamic transitions of the two film structures are significantly different. These measurements quant. demonstrate the strong effects of surface morphol. on the anchoring, order, and dynamics of liq. cryst. thin films.

CC 75-11 (Crystallography and Liquid Crystals)
Section cross-reference(s): 66, 74

IT 40817-08-1, 5CB

RL: PRP (Properties)

(effects of surface morphol. on anchoring and electrooptical dynamics of confined nanoscale 5CB liq. cryst. films bounded by interdigitated electrode array on patterned zinc selenide substrates)

IT 40817-08-1, 5CB

RL: PRP (Properties)

(effects of surface morphol. on anchoring and electrooptical dynamics of confined nanoscale 5CB liq. cryst. films bounded by interdigitated electrode array on patterned zinc selenide substrates)

RN 40817-08-1 HCAPLUS

CN [1,1'-Biphenyl]-4-carbonitrile, 4'-pentyl- (9CI) (CA INDEX NAME)

$$Me-(CH_2)_4$$
 CN

RE.CNT 48 THERE ARE 48 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L26 ANSWER 3 OF 18 HCAPLUS COPYRIGHT 2003 ACS

AN 2002:31094 HCAPLUS

DN 136:93582

TI Liquid crystal composition comprising discotic liquid crystal molecules and alignment promoter

IN Ichihashi, Mitsuyoshi; Kawata, Ken; Takeuchi, Hiroshi; Matsuoka, Koushin

opplicante

PA Fuji Photo Film Co., Ltd., Japan

SO Eur. Pat. Appl., 115 pp.

CODEN: EPXXDW

DT Patent

LA English

FAN.CNT 1

FAN. CNT I								
	PATENT NO.	KIND D	DATE	APPLICATION NO.	DATE			
PI	EP 1170353	A2 2	20020109	EP 2001-115725	20010706			
	EP 1170353	A3 2	20030122					
	R: AT, BE,	CH, DE,	DK, ES, FR,	GB, GR, IT, LI, LU,	, NL, SE, MC, PT,			
	IE, SI,	LT, LV,	FI, RO					
	JP 2002020363	A2 2	20020123	JP 2000-205710	20000706			
	JP 2002038157	A2 2	20020206	JP 2000-220963	20000721			
	US 2002039627	A1 2	20020404	UŞ 2001-899031	20010706			
	JP 2002129162	A2 2	20020509	JP 2001-206337	20010706			
PRAI	JP 2000-205709	A 2	20000706					
	JP 2000-205710	A 2	20000706					
	JP 2000-220963	A 2	20000721					
Λď	MADDAT 126.0250	2						

OS MARPAT 136:93582

AB A liq. crystal compn. comprises liq. crystal mols. and an

alignment promoter. The alignment

promoter is represented by the general formula (Hb-L1Cy1-L2)nAr (Hb = C6-40-aliph., C1-40-aliph. oligosiloxanoxy group; L1 = single bond, divalent linking group comprising alkylene, fluorine-substituted alkylene, -O-, -S-, -CO-, -NR-, -SO2-; L2 = single bond, divalent linking group comprising alkylene, alkenylene, alkynylene, -O-, -S-, -CO-, -NR-, -SO2-; R = H, C1-30-alkyl; Cy1 = divalent arom. or heterocyclic; n = 2 - 5; Ar = n-valent arom.). The object of the present invention is to provide a liq. crystal compn. in which liq. crystal mols. can easily be aligned uniformly. Another object of the invention is to provide an optically anisotropic element in which liq. crystal mols. are uniformly aligned near an interface having no orientation layer.

IC ICM C09K019-56

ICS G02F001-1337

```
74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other
     Reprographic Processes)
     Section cross-reference(s): 75
     liq discotic crystal display alignment promoter compd;
ST
     orientation layer alignment promoter compd liq
     discotic crystal display
IT
     Liquid crystals
        (discotic; liq. crystal compn. comprising discotic and rod-like liq.
        crystal mols. and alignment promoter)
IT
     Liquid crystal displays
        (lig. crystal compn. comprising discotic and rod-like lig. crystal
        mols. and alignment promoter)
IT
     Molecular orientation
        (liq. crystal compn. comprising discotic and rod-like liq. crystal
        mols. and alignment promoter in relation to)
IT
     4236-15-1, Megafac F 104
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (Megafac F-104; in synthesis of alignment promoter
        compd.)
ΙT
     66230-67-9, ZLI 1132
     RL: PEP (Physical, engineering or chemical process); TEM (Technical or
     engineered material use); PROC (Process); USES (Uses)
        (ZLI 1132; liq. crystal compn. comprising discotic and rod-like liq.
        crystal mols. and alignment promoter)
TI
     381233-68-7P 387822-61-9P 387822-68-6P
     387822-70-0P
     RL: PRP (Properties); SPN (Synthetic preparation); TEM (Technical or
     engineered material use); PREP (Preparation); USES (Uses)
        (alignment promoter; liq. crystal compn. comprising
        discotic and rod-like liq. crystal mols. and alignment
        promoter)
IT
     387822-63-1P 387822-66-4P
     RL: SPN (Synthetic preparation); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (alignment promoter; liq. crystal compn. comprising
        discotic and rod-like liq. crystal mols. and alignment
        promoter)
IΤ
     387822-74-4 387822-75-5 387822-77-7
     387822-78-8 387822-79-9
     RL: TEM (Technical or engineered material use); USES (Uses)
        (alignment promoter; liq. crystal compn. comprising
        discotic and rod-like liq. crystal mols. and alignment
        promoter)
IT
     89-55-4, 5-Bromosalicylic acid 92-70-6,
     3-Hydroxy-2-naphthoic acid 100-02-7, p-Nitrophenol, reactions
     108-73-6, 1,3,5-Trihydroxybenzene 108-77-0, Cyanuric chloride
     536-74-3, Phenylacetylene 885-82-5 1095-03-0,
     Phenyl borate 1321-05-7, Bromosalicylic acid 30136-15-3,
     Nitrocatechol
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (in synthesis of alignment promoter compd.)
     323-87-5P, 5-Phenylsalicylic acid 7163-25-9P
     17504-14-2P 37540-59-3P
                              387822-37-9P
     387822-38-0P 387822-39-1P 387822-40-4P
     387822-41-5P 387822-42-6P 387822-43-7P
     387822-44-8P 387822-46-0P 387822-50-6P
     387822-53-9P 387822-55-1P 387822-58-4P
     387822-59-5P
```

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(in synthesis of alignment promoter compd.)

IT 332112-04-6

RL: DEV (Device component use); USES (Uses)

(liq. crystal compn. comprising discotic and rod-like liq. crystal mols. and alignment promoter)

IT 387822-81-3

RL: PEP (Physical, engineering or chemical process); PRP (Properties); TEM (Technical or engineered material use); PROC (Process); USES (Uses)

(liq. crystal compn. comprising discotic and rod-like liq. crystal mols. and alignment promoter)

IT 173071-44-8

RL: PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses)

(liq. crystal compn. comprising discotic and rod-like liq. crystal mols. and alignment promoter)

IT 132694-65-6 339588-79-3

RL: TEM (Technical or engineered material use); USES (Uses) (rod-like liq. crystal; liq. crystal compn. comprising discotic and rod-like liq. crystal mols. and alignment promoter)

IT 66230-67-9, ZLI 1132

RL: PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses)

(ZLI 1132; liq. crystal compn. comprising discotic and rod-like liq. crystal mols. and alignment promoter)

RN 66230-67-9 HCAPLUS

CN [1,1'-Biphenyl]-4-carbonitrile, 4'-(4-pentylcyclohexyl)-, trans-, mixt. with 4-(trans-4-heptylcyclohexyl)benzonitrile, 4-(trans-4-pentylcyclohexyl)benzonitrile and 4-(trans-4-propylcyclohexyl)benzonitrile (9CI) (CA INDEX NAME)

CM 1

CRN 68065-81-6 CMF C24 H29 N

Relative stereochemistry.

CM 2

CRN 61204-03-3 CMF C20 H29 N

Relative stereochemistry.

CM 3

CRN 61204-01-1 CMF C18 H25 N

Relative stereochemistry.

CM 4

CRN 61203-99-4 CMF C16 H21 N

Relative stereochemistry.

IT 381233-68-7P 387822-61-9P 387822-68-6P 387822-70-0P

RL: PRP (Properties); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(alignment promoter; liq. crystal compn. comprising

discotic and rod-like liq. crystal mols. and alignment

promoter)

RN

381233-68-7 HCAPLUS

CN 1-Octanesulfonamide, N,N',N'',N''',N'''',N''''-[1,3,5-triazine-2,4,6-triyltris[imino-4,1,2-benzenetriylbis(oxy-2,1-

ethanediyl)]]hexakis[1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro-N-propyl- (9CI) (CA INDEX NAME)

PAGE 1-B

$$-(CF_2)_7 - CF_3$$

PAGE 2-A

RN 387822-61-9 HCAPLUS

CN 1-Octanesulfonamide, N,N',N''-[1,3,5-triazine-2,4,6-triyltris(imino-4,1-phenyleneoxy-2,1-ethanediyl)]tris[1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro-N-propyl- (9CI) (CA INDEX NAME)

PAGE 1-A

$$F_{3}C-(CF_{2})_{7}-S=0$$

$$n-Pr-N-CH_{2}-CH_{2}-0$$

$$F_{3}C-(CF_{2})_{7}-S=0$$

$$n-Pr-N-CH_{2}-CH_{2}-0$$

$$NH$$

$$NH$$

$$NH$$

PAGE 1-B

$$\begin{array}{c} \text{O} \\ || \\ \text{O} \\ = \text{S} - (\text{CF}_2)_7 - \text{CF}_3 \\ | \\ - \text{CH}_2 - \text{CH}_2 - \text{N} - \text{Pr} - \text{n} \end{array}$$

RN 387822-68-6 HCAPLUS

CN Benzoic acid, 2-[2-[[(heptadecafluorooctyl)sulfonyl]propylamino]ethoxy]-5-(phenylethynyl)-, 1,3,5-benzenetriyl ester (9CI) (CA INDEX NAME)

$$F_{3}C-(CF_{2})_{7}-S=0$$

$$n-Pr-N-CH_{2}-CH_{2}-O$$

$$C=0$$

$$C$$

PAGE 1-B

$$\begin{array}{c}
0 \\
\parallel \\
0 = s - (CF_2)_7 - CF_3 \\
- CH_2 - N - Pr - n
\end{array}$$

 $-(CF_2)_7 - CF_3$

RN CN

387822-70-0 HCAPLUS

2-Naphthalenecarboxylic acid, 3-[2-[[(heptadecafluorooctyl)sulfonyl]propyl amino]ethoxy]-, 1,3,5-benzenetriyl ester (9CI) (CA INDEX NAME)

$$O = S - (CF_2) 7 - CF_3$$

$$O - CH_2 - CH_2 - N - Pr - n$$

$$O - CH_2 - CH_2 - N - Pr - n$$

$$O - CH_2 - CH_2 - N - Pr - n$$

$$O - CH_2 - CH_2 - N - Pr - n$$

$$O - CH_2 - CH_2 - N - Pr - n$$

PAGE 2-A

IT 387822-63-1P 387822-66-4P

RL: SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(alignment promoter; liq. crystal compn. comprising discotic and rod-like liq. crystal mols. and alignment

promoter)

RN

CN

387822-63-1 HCAPLUS

1-Octanesulfonamide, N,N',N''-[1,3,5-triazine-2,4,6-triyltris(imino[1,1'-biphenyl]-3,4-diyloxy-2,1-ethanediyl)]tris[1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro-N-propyl- (9CI) (CA INDEX NAME)

$$F_{3}C-(CF_{2})_{7}-S=0$$

$$n-Pr-N-CH_{2}-CH_{2}-O$$

$$ph$$

$$NH$$

$$Ph$$

$$NH$$

$$NH$$

$$NH$$

$$O$$

$$S$$

$$n-Pr-N-CH_{2}-CH_{2}-O$$

$$O-CH_{2}-CH_{2}-N$$

PAGE 1-B

 $-(cF_2)_7 - cF_3$

— Pr-n

 $\mathcal {U}$

RN 387822-66-4 HCAPLUS

CN [1,1'-Biphenyl]-3-carboxylic acid, 4-[2-[[(heptadecafluorooctyl)sulfonyl]propylamino]ethoxy]-, 1,3,5-benzenetriyl ester (9CI) (CA INDEX NAME)

$$F_{3}C-(CF_{2})_{7}-S=0$$

$$n-Pr-N-CH_{2}-CH_{2}-O$$

$$r_{3}C-(CF_{2})_{7}-S=0$$

$$r_{3}C-(CF_{2})_{7}-S=0$$

$$r_{3}C-(CF_{2})_{7}-S=0$$

$$r_{4}C-CH_{2}-CH_{2}-O$$

$$r_{5}C-CH_{2}-CH_{2}-O$$

$$r_{5}C-CH_{2}-CH_{2}-O$$

$$r_{5}C-CH_{2}-CH_{2}-O$$

$$r_{5}C-CH_{2}-CH_{2}-O$$

$$r_{5}C-CH_{2}-CH_{2}-O$$

PAGE 1-B

IT 387822-74-4 387822-75-5 387822-77-7

387822-78-8 387822-79-9

RL: TEM (Technical or engineered material use); USES (Uses) (alignment promoter; liq. crystal compn. comprising discotic and rod-like liq. crystal mols. and alignment promoter)

X RN CN

N 387822-74-4 HCAPLUS

Benzoic acid, 2-[2-[[(heptadecafluorooctyl)sulfonyl]propylamino]ethoxy]-, 1,3,5-benzenetriyl ester (9CI) (CA INDEX NAME)

RN 387822-75-5 HCAPLUS

CN 1-Octanesulfonamide, N,N',N''-[1,3,5-triazine-2,4,6-triyltris(imino-3,1-phenyleneoxy-2,1-ethanediyl)]tris[1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro-N-propyl- (9CI) (CA INDEX NAME)

 $-(cF_2)_7-cF_3$

 $0 = S - (CF_2)_7 - CF_3 \\
- CH_2 - CH_2 - N - Pr - n$

RN 387822-77-7 HCAPLUS

CN 2-Naphthalenecarboxylic acid, 3-[2-[[(heptadecafluorooctyl)sulfonyl]propyl amino]ethoxy]-, 1,4-phenylene ester (9CI) (CA INDEX NAME)

 $F_{3}C-(CF_{2})_{7}-S=0$ $n-Pr-N-CH_{2}-CH_{2}-O$ $r-Pr-N-CH_{2}-CH_{2}-O$ $r-Pr-N-CH_{2}-CH_{2}-O$ $r-Pr-N-CH_{2}-CH_{2}-O$

RN 387822-78-8 HCAPLUS

CN

2-Naphthalenecarboxylic acid, 3-[2-[[(heptadecafluorooctyl)sulfonyl]propyl amino]ethoxy]-, 1,3-phenylene ester (9CI) (CA INDEX NAME)

RN 387822-79-9 HCAPLUS

CN Benzoic acid, 2-[2-[[(heptadecafluorooctyl)sulfonyl]propylamino]ethoxy]-5-(phenylazo)-, 1,3,5-benzenetriyl ester (9CI) (CA INDEX NAME)

$$F_{3}C-(CF_{2})_{7}-S=0$$

$$n-Pr-N-CH_{2}-CH_{2}-O$$

$$F_{3}C-(CF_{2})_{7}-S=0$$

$$n-Pr-N-CH_{2}-CH_{2}-O$$

$$r_{3}C-(CF_{2})_{7}-S=0$$

$$r_{4}-CH_{2}-CH_{2}-O$$

$$r_{5}-CH_{2}-CH_{2}-O$$

$$r_{5}-CH_{2}-CH_{2}-O$$

$$r_{5}-CH_{2}-CH_{2}-O$$

$$r_{5}-CH_{2}-CH_{2}-O$$

$$r_{5}-CH_{2}-CH_{2}-O$$

PAGE 1-B

$$= s - (CF_2) 7 - CF_3$$

 $= N - Pr - n$

IT 89-55-4, 5-Bromosalicylic acid 92-70-6,

KATHLEEN FULLER EIC 1700/PARKER LAW 308-4290

```
FEELY 09/899031 Page 17
```

3-Hydroxy-2-naphthoic acid 536-74-3, Phenylacetylene 885-82-5 1095-03-0, Phenyl borate 1321-05-7, Bromosalicylic acid RL: RCT (Reactant); RACT (Reactant or reagent) (in synthesis of alignment promoter compd.)

RN 89-55-4 HCAPLUS
CN Benzoic acid, 5-bromo-2-hydroxy- (9CI) (CA INDEX NAME)

RN 92-70-6 HCAPLUS CN 2-Naphthalenecarboxylic acid, 3-hydroxy- (9CI) (CA INDEX NAME)

RN 536-74-3 HCAPLUS CN Benzene, ethynyl- (8CI, 9CI) (CA INDEX NAME)

 $Ph-C \equiv CH$

RN 885-82-5 HCAPLUS CN [1,1'-Biphenyl]-4-ol, 3-nitro- (9CI) (CA INDEX NAME)

RN 1095-03-0 HCAPLUS CN Boric acid (H3BO3), triphenyl ester (8CI, 9CI) (CA INDEX NAME)

RN 1321-05-7 HCAPLUS CN Benzoic acid, bromo-2-hydroxy- (9CI) (CA INDEX NAME)

D1-Br

RN 7163-25-9 HCAPLUS
CN 2-Naphthalenecarboxylic acid, 3-hydroxy-, ethyl ester (9CI) (CA INDEX NAME)

RN 37540-59-3 HCAPLUS

CN Benzoic acid, 5-bromo-2-hydroxy-, ethyl ester (9CI) (CA INDEX NAME)

RN 387822-38-0 HCAPLUS

CN 1-Octanesulfonamide, 1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro-N-[2-(4-nitrophenoxy)ethyl]-N-propyl- (9CI) (CA INDEX NAME)

$$O = S - (CF_2)_7 - CF_3$$

$$O - CH_2 - CH_2 - N - Pr - n$$

$$O_2N$$

RN 387822-39-1 HCAPLUS

CN 1-Octanesulfonamide, N-[2-(4-aminophenoxy)ethyl]1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro-N-propyl- (9CI) (CA INDEX NAME)

$$O = S - (CF_2) 7 - CF_3$$

$$O = CH_2 - CH_2 - N - Pr - n$$

$$H_2N$$

RN 387822-40-4 HCAPLUS

CN 1-Octanesulfonamide, N,N'-[(4-nitro-1,2-phenylene)bis(oxy-2,1-

KATHLEEN FULLER EIC 1700/PARKER LAW 308-4290

ethanediyl)]bis[1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro-N-propyl-(9CI) (CA INDEX NAME)

$$F_{3}C-(CF_{2})_{7}-S=0$$
 $n-Pr-N-CH_{2}-CH_{2}-0$
 $O=S-(CF_{2})_{7}-CF_{3}$
 $O-CH_{2}-CH_{2}-N-Pr-n$

RN 387822-41-5 HCAPLUS

CN 1-Octanesulfonamide, N,N'-[(4-amino-1,2-phenylene)bis(oxy-2,1-ethanediyl)]bis[1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro-N-propyl-(9CI) (CA INDEX NAME)

$$F_3C-(CF_2)_7-S=0$$
 $n-Pr-N-CH_2-CH_2-0$
 $O=S-(CF_2)_7-CF_3$
 $O-CH_2-CH_2-N-Pr-n$

RN 387822-42-6 HCAPLUS

CN 2-Naphthalenecarboxylic acid, 3-[2-[[(heptadecafluorooctyl)sulfonyl]propyl amino]ethoxy]-, ethyl ester (9CI) (CA INDEX NAME)

RN 387822-43-7 HCAPLUS

CN 2-Naphthalenecarboxylic acid, 3-[2-[[(heptadecafluorooctyl)sulfonyl]propyl amino]ethoxy]- (9CI) (CA INDEX NAME)

RN 387822-44-8 HCAPLUS

CN Benzoic acid, 5-bromo-2-[2-[[(heptadecafluorooctyl)sulfonyl]propylamino]et hoxy]-, ethyl ester (9CI) (CA INDEX NAME)

$$F_{3}C-(CF_{2})_{7}-S=0$$

$$n-Pr-N-CH_{2}-CH_{2}-0$$

$$C-OET$$

RN 387822-46-0 HCAPLUS

CN Benzoic acid, 2-[2-[[(heptadecafluorooctyl)sulfonyl]propylamino]ethoxy]-5-(phenylethynyl)-, ethyl ester (9CI) (CA INDEX NAME)

$$F_{3}C-(CF_{2})_{7}-S=0$$

$$n-Pr-N-CH_{2}-CH_{2}-O$$

$$C=C-Ph$$

$$C-OEt$$

RN 387822-50-6 HCAPLUS

CN Benzoic acid, 2-[2-[[(heptadecafluorooctyl)sulfonyl]propylamino]ethoxy]-5-(phenylethynyl)- (9CI) (CA INDEX NAME)

$$C = C - Ph$$

$$F_3C - (CF_2)_7 - S = O$$

$$n - Pr - N - CH_2 - CH_2 - O$$

$$CO_2H$$

RN 387822-53-9 HCAPLUS

CN 1-Octanesulfonamide, 1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro-N-

[2-[(3-nitro[1,1'-biphenyl]-4-yl)oxy]ethyl]-N-propyl- (9CI) (CA INDEX NAME)

$$O = S - (CF_2)_7 - CF_3$$

$$O = S - (CF_2)_7 - CF_3$$

$$O = CH_2 - CH_2 - N - Pr - n$$

$$Ph$$

RN 387822-55-1 HCAPLUS

CN 1-Octanesulfonamide, N-[2-[(3-amino[1,1'-biphenyl]-4-yl)oxy]ethyl]1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro-N-propyl- (9CI) (CA INDEX NAME)

$$O = S - (CF_2)_7 - CF_3$$

$$O = S - (CF_2)_7 - CF_3$$

$$O = CH_2 - CH_2 - N - Pr - n$$
Ph

RN 387822-58-4 HCAPLUS

CN [1,1'-Biphenyl]-3-carboxylic acid, 4-[2-[[(heptadecafluorooctyl)sulfonyl]propylamino]ethoxy]-, ethyl ester (9CI) (CA INDEX NAME)

RN 387822-59-5 HCAPLUS

CN [1,1'-Biphenyl]-3-carboxylic acid, 4-[2-[[(heptadecafluorooctyl)sulfonyl]propylamino]ethoxy]- (9CI) (CA INDEX NAME)

IT 332112-04-6

RL: DEV (Device component use); USES (Uses)
(liq. crystal compn. comprising discotic and rod-like liq. crystal mols. and alignment promoter)

RN 332112-04-6 HCAPLUS

CN 2-Propenoic acid, 3-[4-[4-[(1-oxo-2-propenyl)oxy]butoxy]phenyl]-, 2,3,6,7,10,11-triphenylenehexayl ester (9CI) (CA INDEX NAME)

PAGE 1-A

$$H_2C = CH - C - O - (CH_2)_4 - O$$
 CH
 CH

PAGE 2-B

PAGE 3-A
H₂C== CH-C-O-(CH₂)₄-O

IT 387822-81-3

RL: PEP (Physical, engineering or chemical process); PRP (Properties); TEM (Technical or engineered material use); PROC (Process); USES (Uses) (liq. crystal compn. comprising discotic and rod-like liq. crystal mols. and alignment promoter)

RN 387822-81-3 HCAPLUS

CN D-Glucitol, 1,4:3,6-dianhydro-, bis[4-[4-[(1-oxo-2-propenyl)oxy]butoxy]benzoate], polymer with 2,6-naphthalenediyl bis[4-[4-[(1-oxo-2-propenyl)oxy]butoxy]benzoate] and 1,4-phenylene bis[4-[4-[(1-oxo-2-propenyl)oxy]butoxy]benzoate] (9CI) (CA INDEX NAME)

CM 1

CRN 339588-79-3 CMF C38 H36 O10

PAGE 1-A

$$H_2C = CH - C - O - (CH_2)_4 - O$$

PAGE 1-B

CM 2

CRN 250230-59-2 CMF C34 H38 O12

Absolute stereochemistry.

PAGE 1-A

$$H_2C$$
 O
 $CCH_2)$ A
 O
 R
 R
 R
 S
 O
 H

PAGE 1-B

CM 3

CRN 132694-65-6 CMF C34 H34 O10

PAGE 1-B

$${\rm CH_2}$$
) ${\rm _4-o-C-CH}$

IT 173071-44-8

RL: PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses)

(liq. crystal compn. comprising discotic and rod-like liq. crystal mols. and alignment promoter)

RN 173071-44-8 HCAPLUS

CN Benzoic acid, 4-[[6-[(1-oxo-2-propenyl)oxy]hexyl]oxy]-, 2,3,6,7,10,11-triphenylenehexayl ester (9CI) (CA INDEX NAME)

PAGE 1-A

$$H_2C = CH - C - O - (CH_2)_6 - O$$
 $C = O$
 C

PAGE 2-A

IT 132694-65-6 339588-79-3

RL: TEM (Technical or engineered material use); USES (Uses) (rod-like liq. crystal; liq. crystal compn. comprising discotic and rod-like liq. crystal mols. and alignment promoter)

RN 132694-65-6 HCAPLUS

CN Benzoic acid, 4-[4-[(1-oxo-2-propenyl)oxy]butoxy]-, 1,4-phenylene ester (9CI) (CA INDEX NAME)

PAGE 1-B

RN 339588-79-3 HCAPLUS

CN Benzoic acid, 4-[4-[(1-oxo-2-propenyl)oxy]butoxy]-, 2,6-naphthalenediyl ester (9CI) (CA INDEX NAME)

PAGE 1-A

$$_{\text{H}_2\text{C}} = _{\text{CH}-\text{C}-\text{O}-\text{(CH}_2)}^{\text{O}} _{\text{4}-\text{O}} = _{\text{C}-\text{O}}^{\text{O}} = _{\text{C}-\text{O}}^{\text{O}}$$

PAGE 1-B

L26 ANSWER 4 OF 18 HCAPLUS COPYRIGHT 2003 ACS

AN 2001:153786 HCAPLUS

DN 134:373964

TI Allyl p-fluor cinnamate grafted polysiloxane photoalignment films polymerized under linear polarized UV light

AU Liu, J.; Liang, X.; Tang, H.; Xu, S.; Gao, H.

CS Department of Chemistry, Tsinghua University, Beijing, 100084, Peop. Rep.

SO Thin Solid Films (2001), 384(2), 212-214 CODEN: THSFAP; ISSN: 0040-6090

PB Elsevier Science S.A.

DT Journal

LA English

AB Polysiloxane allyl p-fluor cinnamate (PSAFC) photopolymer films promote uniform alignment of adjacent liq. crystals (LC) upon photopolymn. by linear polarized UV light. Such films show excellent thermal stability with considerable dichroism after undergoing linear polarized UV-induced cycloaddn. reaction. There is no clear difference between the at. force microscope (AFM) images of the films before and after the exposure, which shows that there is no clear morphol. anisotropy on the exposed film.

CC 74-12 (Radiation Chemistry, Photochemistry, and Photographic and Other

Reprographic Processes)

IT 270907-93-2D, graft copolymer with polysiloxane
RL: TEM (Technical or engineered material use); USES (Uses)
(thermally stable allyl p-fluorocinnamate-grafted polysiloxane photoalignment film for uniform alignment of liq. crystals upon photopolymn. under polarized UV light)

RN 270907-93-2 HCAPLUS

CN 2-Propenoic acid, 3-(4-fluorophenyl)-, 2-propenyl ester, (2E)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

RE.CNT 10 THERE ARE 10 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L26 ANSWER 5 OF 18 HCAPLUS COPYRIGHT 2003 ACS

AN 2000:219340 HCAPLUS

DN 132:315714

TI Modeling the Interface Region of Command Surface 2. Spectroscopic Evaluations of Azobenzene/Liquid Crystal Hybrid Langmuir-Blodgett Films under Illumination

AU Ubukata, Takashi; Seki, Takahiro; Morino, Shin'ya; Ichimura, Kunihiro

CS Photofunctional Chemistry Division Research Laboratory of Resources Utilization, Tokyo Institute of Technology, Yokohama, 226-8503, Japan

SO Journal of Physical Chemistry B (2000), 104(17), 4148-4154 CODEN: JPCBFK; ISSN: 1089-5647

PB American Chemical Society

DT Journal

LA English

AB Langmuir-Blodgett (LB) films composed of the mixt. of an amphiphilic polymer contg. azobenzene (Az) side chain (6Az10-PVA) and 4'-pentyl-4-cyanobiphenyl (5CB) were prepd. to mimic the 2-dimensional contacting region of the LC/Az interface of the command surface which

photochem. switches the LC alignment. UV-visible absorption and FTIR spectroscopic measurements were carried out under illumination. These procedures allowed sep. and simultaneous evaluations of the static state and dynamic mol. motions of both Az and LC mols., which probably reflect the initial triggering step of the domino-mode response of LC. The spectroscopic data indicated the induction of reversible perpendicular/tilt orientational changes of both the Az side chain and 5CB mol. upon alternative irradn. of 365 and 436 nm light. Thus, 6Az10-PVA/5CB hybrid LB film can be regarded as a satisfactory interface model of a command surface that **promotes** the homeotropic/planer alignment switching. From the time courses of the photoisomerization of Az and the orientational change, the mol. tilt is not governed only by the trans/cis ratio of Az unit, but is strongly process-dependent (forward or back process), indicative of involvement of strong mol. cooperativity. The validity and limitation of the LC research using this model system are also discussed.

CC 74-1 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

Section cross-reference(s): 66, 73

IT 40817-08-1

RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process)

(modeling interface region of photochem. switch command surface with spectroscopic evaluations of azobenzene/liq. crystal hybrid Langmuir-Blodgett films under illumination)

IT 52364-71-3, 50CB 151752-94-2

RL: PRP (Properties)

(modeling interface region of photochem. switch command surface with spectroscopic evaluations of azobenzene/liq. crystal hybrid Langmuir-Blodgett films under illumination)

IT 40817-08-1

CN

RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process)

(modeling interface region of photochem. switch command surface with spectroscopic evaluations of azobenzene/liq. crystal hybrid Langmuir-Blodgett films under illumination)

RN 40817-08-1 HCAPLUS

[1,1'-Biphenyl]-4-carbonitrile, 4'-pentyl- (9CI) (CA INDEX NAME)

$$Me-(CH_2)_4$$
 CN

IT 52364-71-3, 50CB 151752-94-2

RL: PRP (Properties)

(modeling interface region of photochem. switch command surface with spectroscopic evaluations of azobenzene/liq. crystal hybrid Langmuir-Blodgett films under illumination)

RN 52364-71-3 HCAPLUS

CN [1,1'-Biphenyl]-4-carbonitrile, 4'-(pentyloxy)- (9CI) (CA INDEX NAME)

RN 151752-94-2 HCAPLUS

CN Undecanoic acid, 11-[4-[(4-hexylphenyl)azo]phenoxy]-, ethenyl ester, polymer with ethenol (9CI) (CA INDEX NAME)

CM 1

CRN 151752-93-1 CMF C31 H44 N2 O3

$$_{\rm H_2C}=_{\rm CH-O-C-(CH_2)_{10}-O}$$
 $_{\rm N=N-M=N-M=0}$ (CH₂)₅-Me

CM 2

CRN 557-75-5 CMF C2 H4 O

 $H_2C = CH - OH$

RE.CNT 32 THERE ARE 32 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L26 ANSWER 6 OF 18 HCAPLUS COPYRIGHT 2003 ACS

AN 1999:119297 HCAPLUS

DN 130:230859

TI Control of the bias tilt angles in nematic liquid crystals

AU Yablonskii, S. V.; Nakayama, K.; Okazaki, S.; Ozaki, M.; Yoshino, K.; Palto, S. P.; Baranovich, M. Yu.; Michailov, A. S.

CS Faculty of Engineering, Department of Electronic Engineering, Osaka University, 2-1 Ymada-Oka, Suita, Osaka, 565-0871, Japan

SO Journal of Applied Physics (1999), 85(5), 2556-2561 CODEN: JAPIAU; ISSN: 0021-8979

PB American Institute of Physics

DT Journal

LA English

AB The pretilt angle controlled by elec. field was studied by the modulation ellipsometry technique. The easy direction of compensated nematic liq. crystals was controlled by surface flexoelec. torque created by the linear coupling of the director deformation and elec. field. The weak anchoring energy necessary for the occurrence of flexoelec. distortion was produced by unidirectional rubbing of the clean In-Sn-oxide covered glasses with a cotton cloth. The pretilt angle was measured as a function of elec. field. Long relaxation times of the optical response (hundreds of seconds) were obsd. The rubbed thin polyvinyl alc. and polyimide

aligning layers promote strong anchoring energy (>0.5 erg/cm2) preventing any deviation of pretilt angle and, consequently, to suppress the optical response. The probable applications of the obtained results are discussed.

CC 76-13 (Electric Phenomena)
 Section cross-reference(s): 75

IT 11106-72-2, p-Butyl-p'-heptanoyloxyazoxybenzene 38690-76-5
97402-82-9, MBBA

RL: PEP (Physical, engineering or chemical process); PRP (Properties); TEM (Technical or engineered material use); PROC (Process); USES (Uses) (control of bias tilt angles in nematic liq. crystals by elec. fields)

IT 38690-76-5 97402-82-9, MBBA
RL: PEP (Physical, engineering or chemical process); PRP (Properties); TEM (Technical or engineered material use); PROC (Process); USES (Uses) (control of bias tilt angles in nematic liq. crystals by elec. fields)

RN 38690-76-5 HCAPLUS CN Benzoic acid, 4-heptyl-, 4-cyanophenyl ester (9CI) (CA INDEX NAME)

RN 97402-82-9 HCAPLUS

CN Benzenamine, 4-butyl-N-[(4-methoxyphenyl)methylene]-, [N(E)]- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

RE.CNT 15 THERE ARE 15 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L26 ANSWER 7 OF 18 HCAPLUS COPYRIGHT 2003 ACS

AN 1998:441902 HCAPLUS

DN 129:149554

TI Morphological investigation of polyvinyl-4-methoxy cinnamate photopolymer thin and ultrathin films under linear photopolymerization

AU Rajesh, K.; Ram, M. K.; Jain, S. C.; Samanta, S. B.; Narliker, A. V. CS Display Devices Group, National Physical Laboratory, New Delhi, India

SO Thin Solid Films (1998), 325(1,2), 251-253

CODEN: THSFAP; ISSN: 0040-6090

PB Elsevier Science S.A.

DT Journal

LA English

AB The characteristics of the surface of poly(vinyl-4-methoxy cinnamate) (PVMC) photopolymer film which **promotes** the uniform

alignment in adjacent liq. crystals upon photopolymn. by linear polarized UV light have been investigated. Under photopolymn. the surface morphol. of the polymer film changes and the polymer orients normal to the polarization direction of the UV light. The results confirm the microscopic model of the linear photopolymd. induced unidirectional order as reported by Schadt et al. (M. Schadt, K. Schmitt, V. Koznikov, V. Chignirov, Jpn. J. Appl. Phys. 31 (1992) 2155).

CC 37-5 (Plastics Manufacture and Processing)

IT **61204-01-1**, PCH-5

RL: NUU (Other use, unclassified); USES (Uses)

(liq. crystal; morphol. of poly(vinyl-4-methoxy cinnamate) thin and ultrathin films after polymn. with linear polarized UV light at sub-micron level in presence of)

IT 32732-28-8, Poly(vinyl-4-methoxy cinnamate)

RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process)

(morphol. of poly(vinyl-4-methoxy cinnamate) thin and ultrathin films after polymn. with linear polarized UV light at sub-micron level)

IT **61204-01-1**, PCH-5

RL: NUU (Other use, unclassified); USES (Uses)

(liq. crystal; morphol. of poly(vinyl-4-methoxy cinnamate) thin and ultrathin films after polymn. with linear polarized UV light at sub-micron level in presence of)

RN 61204-01-1 HCAPLUS

CN Benzonitrile, 4-(trans-4-pentylcyclohexyl)- (9CI) (CA INDEX NAME)

Relative stereochemistry.

IT 32732-28-8, Poly(vinyl-4-methoxy cinnamate)

RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process)

(morphol. of poly(vinyl-4-methoxy cinnamate) thin and ultrathin films after polymn. with linear polarized UV light at sub-micron level)

RN 32732-28-8 HCAPLUS

CM 1

CRN 10604-64-5 CMF C12 H12 O3

RE.CNT 8 THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L26 ANSWER 8 OF 18 HCAPLUS COPYRIGHT 2003 ACS

AN 1997:498912 HCAPLUS

DN 127:240913

TI Alignment of liquid crystal on poly(vinyl cinnamate) photopolymer and anchoring direction

AU Rajesh, Kumar; Masuda, Shin; Yamaguchi, Rumiko; Sato, Susumu

CS Department of Electrical and Electronic Engineering, Akita University, Akita, 010, Japan

Japanese Journal of Applied Physics, Part 1: Regular Papers, Short Notes & Review Papers (1997), 36(7A), 4404-4408 CODEN: JAPNDE; ISSN: 0021-4922

PB Japanese Journal of Applied Physics

DT Journal

LA English

AB A poly(vinyl cinnamate) photopolymer under irradn. of linearly polarized UV light acts as an aligning agent inducing orientation in adjacent liq. crystals. We have promoted the homogeneous as well as homeotropic alignment of liq. crystal using poly(vinyl cinnamate) as an aligning agent, under different exptl. conditions. The anticipated orientation mechanism for the promotion of homeotropic alignment in the case of a liq. crystal/photopolymer composite system is discussed. From our observations we infer that in a liq. crystal/photopolymer composite system with a change of curing condition, the boundary surface condition at the substrates changes, which in turn affects the mode of alignment.

CC 74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

Section cross-reference(s): 75
TT 26227-73-6, p-Methoxybenzylidene

IT 26227-73-6, p-Methoxybenzylidene-p-n-butylaniline 32732-28-8, Poly(vinyl-4-methoxy cinnamate) 40817-08-1, K15 63748-28-7, E7 (Liquid crystal) 148880-66-4, RDN-91207 RL: DEV (Device component use); USES (Uses)

(poly(vinyl-methoxy cinnamate) alignment layer for liq. crystals)
26227-73-6, p-Methoxybenzylidene-p-n-butylaniline

IT 26227-73-6, p-Methoxybenzylidene-p-n-butylaniline
32732-28-8, Poly(vinyl-4-methoxy cinnamate) 40817-08-1,
K15 63748-28-7, E7 (Liquid crystal)

RL: DEV (Device component use); USES (Uses)

(poly(vinyl-methoxy cinnamate) alignment layer for liq. crystals)

RN 26227-73-6 HCAPLUS

CN Benzenamine, 4-butyl-N-[(4-methoxyphenyl)methylene]- (9CI) (CA INDEX NAME)

RN 32732-28-8 HCAPLUS

CN 2-Propenoic acid, 3-(4-methoxyphenyl)-, ethenyl ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 10604-64-5 CMF C12 H12 O3

$$\begin{array}{c} \text{CH} \longrightarrow \text{CH} \longrightarrow \text{CH} \longrightarrow \text{CH} \longrightarrow \text{CH} \longrightarrow \text{CH}_2 \\ \\ \text{MeO} \end{array}$$

RN 40817-08-1 HCAPLUS

CN [1,1'-Biphenyl]-4-carbonitrile, 4'-pentyl- (9CI) (CA INDEX NAME)

$$Me^{-(CH_2)}\frac{1}{4}$$
 CN

RN 63748-28-7 HCAPLUS

CN [1,1':4',1''-Terphenyl]-4-carbonitrile, 4''-pentyl-, mixt. with 4'-heptyl[1,1'-biphenyl]-4-carbonitrile, 4'-(octyloxy)[1,1'-biphenyl]-4-carbonitrile and 4'-pentyl[1,1'-biphenyl]-4-carbonitrile (9CI) (CA INDEX NAME)

CM 1

CRN 54211-46-0 CMF C24 H23 N

CM 2

CRN 52364-73-5

CMF C21 H25 N O

CM3

CRN 41122-71-8 CMF C20 H23 N

CM

CRN 40817-08-1 CMF C18 H19 N

L26 ANSWER 9 OF 18 HCAPLUS COPYRIGHT 2003 ACS

AN 1995:693143 HCAPLUS

123:70500 DN

Cholesteric liquid crystal devices TI

Rosenblatt, Charles; Fisch, Michael R.; Crandall, Karl A.; Petschek, Rolfe IN

Case Western Reserve University, USA PA

PCT Int. Appl., 27 pp. SO

CODEN: PIXXD2

DTPatent

LA English

FAN.							ADDITIONAL NO. DA	TD	
	PAI	ENT NO.		KIND	DATE		APPLICATION NO. DA	re	
PI	WO	9500879		A1	19950105		WO 1994-US6835 19	940616	
		W: CN,							
		RW: AT,	BE,	CH, DE,	, DK, ES,	FR,	GB, GR, IE, IT, LU, M	C, NL, PT,	, SE
	US	5477358		A	19951219		US 1993-81009 19	930621	
	CN	1125986		A	19960703		CN 1994-192535 19	940616	
	CN	1040582		В	19981104				
	ΕP	737326		A1	19961016		EP 1994-921312 19	940616	
		R: AT,	BE,	CH, DE	, DK, ES,	FR,	GB, GR, IE, IT, LI, L	U, MC, NL	, PT, SE
	JΡ	08511881		Т2	19961210		JP 1994-502961 19	940616	

US 5602662 A 19970211 US 1995-389316 19950216

PRAI US 1993-81009 19930621 WO 1994-US6835 19940616

AB A liq. crystal cell is described where the cell walls are treated to **promote** homeotropic alignment of a chiral nematic liq. crystal material and where the liq.-crystal material has a neg. dielec.

anisotropy and includes sufficiently low amt. of chiral material to enable the lig.-crystal director to homeotropic align in the absence of a field.

IC ICM G02F001-1337

CC 74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

IT 165179-06-6 165179-07-7

RL: DEV (Device component use); USES (Uses)

(liq.-crystal device with treated wall for homeotropic alignment)

IT 165179-06-6 165179-07-7

RL: DEV (Device component use); USES (Uses)

(liq.-crystal device with treated wall for homeotropic alignment)

RN 165179-06-6 HCAPLUS

CN [1,1'-Biphenyl]-4-carbonitrile, 4'-(2-methylbutyl)-, (S)-, mixt. with ZLI 2806 (9CI) (CA INDEX NAME)

CM 1

CRN 113782-31-3

CMF Unspecified

CCI MAN

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

CM 2

CRN 63799-11-1

CMF C18 H19 N

Absolute stereochemistry. Rotation (+).

RN 165179-07-7 HCAPLUS

CN [1,1'-Biphenyl]-4-carbonitrile, 4'-(2-methylbutyl)-, (S)-, mixt. with ZLI 4330 (9CI) (CA INDEX NAME)

CM 1

CRN 137545-99-4

CMF Unspecified

CCI MAN

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

CM 2

CRN 63799-11-1 CMF C18 H19 N

Absolute stereochemistry. Rotation (+).

L26 ANSWER 10 OF 18 HCAPLUS COPYRIGHT 2003 ACS

AN 1995:683049 HCAPLUS

DN 123:57946

TI Langmuir-Blodgett Films of Amphiphilic Polysilanes Bearing a Pendant Ammonium Moiety

AU Seki, Takahiro; Tanigaki, Nobutaka; Yase, Kiyoshi; Kaito, Akira; Tamaki, Takashi; Ueno, Katsuhiko; Tanaka, Yuji

CS Research Laboratory of Resources Utilization, Tokyo Institute of Technology, Yokohama, 226, Japan

SO Macromolecules (1995), 28(16), 5609-17 CODEN: MAMOBX; ISSN: 0024-9297

PB American Chemical Society

DT Journal

LA English

Eight homologous amphiphilic polysilanes bearing an ammonium moiety in the AΒ side substituent were synthesized. The spreading behavior at the air-water interface of these polysilanes, mol. film fabrication by the Langmuir-Blodgett (LB) technique, and their UV absorption properties and structural features were investigated. The UV absorption spectrum of the transferred LB films was dependent on the deposition condition, such as the existence of a hydrophobic counteranion in the substrate and the magnitude of mech. compression by the moving barrier. Such spectral changes should be coupled with conformational changes, i.e., trans/gauche populational changes of the Si backbone modified at the air-water interface. A related conformational modification was achieved when the polysilane monolayer was mixed with stearic acid as the lateral spacer. Polarized UV spectroscopy revealed that the Si backbone is preferentially oriented along the dipping direction possibly due to flow orientation on the water surface. This orientational order depended strongly on the mol. structure of the polysilane and the deposition no. The hydrocarbon side chain, the layer structure, and the morphol. of the LB films were evaluated by Fourier transform IR spectroscopy, transmittance electron microscopy, and X-ray reflectometry. These measurements put forth a view that the LB films are composed of periodical double layers having a homogeneous and amorphous character. Furthermore, a multilayered LB film having the aligned Si backbone promoted homogeneous

alignment of a nematic liq. crystal in the dipping direction.

CC 38-3 (Plastics Fabrication and Uses) Section cross-reference(s): 36, 37

IT 112-18-5DP, N,N-Dimethyldodecylamine, reaction products with

chloromethylated polysilanes 124-28-7DP, N,N-Dimethyloctadecylamine, reaction products with chloromethylated polysilanes 4088-22-6DP, N-Methyldioctadecylamine, reaction products with chloromethylated 7378-99-6DP, N,N-Dimethyloctylamine, reaction products with polysilanes chloromethylated polysilanes 31324-77-3DP, Dichloromethylphenylsilane homopolymer, chloromethylated, reaction products with tertiary amines 76188-55-1DP, Dichloromethylphenylsilane homopolymer, sru, chloromethylated, reaction products with tertiary amines 113925-33-0DP, Dichlorodihexylsilane-Dichloromethylphenylsilane copolymer, chloromethylated, reaction products with tertiary amines RL: PRP (Properties); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (prepn. and characterization of Langmuir-Blodgett films of amphiphilic polysilanes bearing pendant ammonium moiety)

31324-77-3DP, Dichloromethylphenylsilane homopolymer, chloromethylated, reaction products with tertiary amines 76188-55-1DP, Dichloromethylphenylsilane homopolymer, sru, chloromethylated, reaction products with tertiary amines 113925-33-0DP, Dichlorodihexylsilane-Dichloromethylphenylsilane copolymer, chloromethylated, reaction products with tertiary amines RL: PRP (Properties); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(prepn. and characterization of Langmuir-Blodgett films of amphiphilic polysilanes bearing pendant ammonium moiety)

RN 31324-77-3 HCAPLUS

CN Silane, dichloromethylphenyl-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 149-74-6 CMF C7 H8 Cl2 Si

IT

RN 76188-55-1 HCAPLUS CN Poly(methylphenylsilylene) (9CI) (CA INDEX NAME)

RN 113925-33-0 HCAPLUS
CN Silane, dichlorodihexyl-, polymer with dichloromethylphenylsilane (9CI) (CA INDEX NAME)

CM 1

CRN 18204-93-8 CMF C12 H26 C12 Si

$$\begin{array}{c} & \text{Cl} \\ | \\ \text{Me- (CH}_2) \, 5 - \text{Si- (CH}_2) \, 5 - \text{Me} \\ | \\ \text{Cl} \end{array}$$

CM 2

CRN 149-74-6 CMF C7 H8 Cl2 Si

L26 ANSWER 11 OF 18 HCAPLUS COPYRIGHT 2003 ACS

AN 1995:241607 HCAPLUS

DN 122:202129

TI A permittive effect on the threshold behavior at low frequencies and the drift of charge carriers with a liquid-crystalline system of cyanophenylcyclohexanes

AU Oh-E, M.; Kondo, K.; Kando, Y.

CS Hitachi Research Laboratory, Hitachi, Ltd., Ibaraki-ken, 319-12, Japan Molecular Crystals and Liquid Crystals Science and Technology, Section A: Molecular Crystals and Liquid Crystals (1994), 250, 51-62 CODEN: MCLCE9; ISSN: 1058-725X

PB Gordon & Breach

DT Journal

LA English

Measurements are reported for threshold voltage dependence on the AΒ frequency of a nematic liq.-cryst. system composed of cyanophenylcyclohexanes (PCHs). The PCH lig. crystal having a strong dipole only at the terminal position of the mols. showed a frequency independent of threshold voltage at low frequencies. In their mixt., increasing permittivity affected the threshold behavior, finally causing a drop in threshold voltage at low frequencies. The difference in systems with and without the drop in threshold voltage is related to the drift of charge carriers. Relaxation time measurements of transmittance against d.c. bias revealed a tendency for the drift of charge carriers in the liq. crystal layer. The liq.-cryst. system composed of PCHs showed relatively slower relaxation than the system with increased permittivity due to addn. of certain dopants. In the migration of charge carriers, a model is proposed considering the changes of permittivity. In this model the assocn. structure of the liq. crystal mols. is altered by increasing permittivity. Thus the drift of charge carriers, which is related to the formation of interfacial polarization between the liq. crystal and

alignment film layers, is promoted by loss of the mol.
assocn. structure.

CC 76-1 (Electric Phenomena)

Section cross-reference(s): 74, 75

IT 122402-85-1 125497-50-9

RL: NUU (Other use, unclassified); USES (Uses)
(liq. crystal orienting substrate; threshold voltage dependence on frequency of nematic liq.-cryst. system composed of cyanophenylcyclohexanes)

IT 62439-33-2 74240-64-5 80944-44-1 86776-50-3 92118-82-6 161860-55-5D, alkyl

deriv.

RL: MOA (Modifier or additive use); USES (Uses) (threshold voltage dependence on frequency of nematic liq.-cryst. system composed of cyanophenylcyclohexanes)

IT 122402-85-1 125497-50-9

RL: NUU (Other use, unclassified); USES (Uses)
(liq. crystal orienting substrate; threshold voltage dependence on frequency of nematic liq.-cryst. system composed of cyanophenylcyclohexanes)

RN 122402-85-1 HCAPLUS

CN Poly[(octahydro-1,3,4,6-tetraoxocyclobuta[1,2-c:3,4-c']dipyrrole-2,5-diyl)1,4-phenyleneoxy-1,4-phenylene[2,2,2-trifluoro-1(trifluoromethyl)ethylidene]-1,4-phenyleneoxy-1,4-phenylene] (9CI) (CA
INDEX NAME)

PAGE 1-A

PAGE 1-B

n

RN 125497-50-9 HCAPLUS

CN Cyclobuta[1,2-c:3,4-c']difurantetrone, tetrahydro-, polymer with 4,4'-[[2,2,2-trifluoro-1-(trifluoromethyl)ethylidene]bis(4,1-phenyleneoxy)]bis[benzenamine] (9CI) (CA INDEX NAME)

CM 1

CRN 69563-88-8 CMF C27 H20 F6 N2 O2

CM 2

CRN 4415-87-6 CMF C8 H4 O6

IT 62439-33-2 74240-64-5 80944-44-1 86776-50-3 92118-82-6 161860-55-5D, alkyl deriv.

RL: MOA (Modifier or additive use); USES (Uses) (threshold voltage dependence on frequency of nematic liq.-cryst. system composed of cyanophenylcyclohexanes)

RN 62439-33-2 HCAPLUS

CN Cyclohexanecarboxylic acid, 4-propyl-, 4-cyanophenyl ester, trans- (9CI) (CA INDEX NAME)

Relative stereochemistry.

RN 74240-64-5 HCAPLUS

CN Benzonitrile, 4-(trans-5-propyl-1,3-dioxan-2-yl)- (9CI) (CA INDEX NAME)

Relative stereochemistry.

RN 80944-44-1 HCAPLUS

CN Benzene, 1-ethoxy-4-(trans-4-propylcyclohexyl)- (9CI) (CA INDEX NAME)

Relative stereochemistry.

RN 86776-50-3 HCAPLUS

CN Benzoic acid, 4-ethyl-, 4-cyano-3-fluorophenyl ester (9CI) (CA INDEX NAME)

RN 92118-82-6 HCAPLUS

CN Benzoic acid, 4-(trans-4-propylcyclohexyl)-, 4-cyano-3-fluorophenyl ester (9CI) (CA INDEX NAME)

Relative stereochemistry.

RN 161860-55-5 HCAPLUS

CN Cyclohexanecarboxylic acid, 2,3-difluorophenyl ester (9CI) (CA INDEX NAME)

L26 ANSWER 12 OF 18 HCAPLUS COPYRIGHT 2003 ACS

AN 1990:593306 HCAPLUS

DN 113:193306

TI Fibers from flexible liquid crystal main-chain polymers. II. Functional copolymers based on the 4,4'-dihydroxy-2,2'-dimethylazoxybenzene mesogen and spacers based on 2-dodecenedioic and nonanedioic acids

AU Lin, C. H.; Maeda, M.; Blumstein, A.

CS Dep. Chem., Univ. Lowell, Lowell, MA, 01854, USA

SO Journal of Applied Polymer Science (1990), 41(5-6), 1009-22 CODEN: JAPNAB; ISSN: 0021-8995

DT Journal

LA English

AΒ Fibers spun from liq.-cryst. solns. or melts are characterized by a high degree of chain alignment which promotes high values of tenacity and tensile modulus in the direction of fiber axis but lower values in the transversal direction. Structural modifications are introduced into the flexible moiety and the mesogenic group to alleviate this shortcoming. The incorporation of a spacer based on 2-dodecenedioic acid into a thermotropic liq.-cryst. polyester was studied. A phase diagram was established to optimize the compn. of such copolyesters. significant increase in mech. properties on crosslinking through UV irradn. was obsd. A remarkable property of such systems was the retention of mol. chain order in spite of temps. in excess of the glass transition temp. This is valid for both crosslinked and uncrosslinked systems. crystn. on annealing of a no. of copolymer compns. was remarkable because it occurred in spite of a certain mismatch in length (.apprx.2 .ANG.) and a difference in the structure of the flexible spacer.

CC 40-3 (Textiles and Fibers)

IT 124895-98-3

RL: USES (Uses)

(fiber, mech. properties of, UV crosslinking effect on)

IT 82851-48-7 **130141-00-3 130141-01-4** 130174-07-1

RL: USES (Uses)

(thermal transition temps. of liq. cryst., fiber spinning in relation to)

IT 124895-98-3

RL: USES (Uses)

(fiber, mech. properties of, UV crosslinking effect on)

RN 124895-98-3 HCAPLUS

CN 2-Dodecenedioic acid, (E)-, polymer with (Z)-4,4'-azoxybis[3-methylphenol] and nonanedioic acid (9CI) (CA INDEX NAME)

CM 1

CRN 78992-83-3 CMF C14 H14 N2 O3

Double bond geometry as shown.

CM 2

CRN 6402-36-4 CMF C12 H20 O4

Double bond geometry as shown.

CM 3

CRN 123-99-9 CMF C9 H16 O4

 $HO_2C-(CH_2)_7-CO_2H$

IT 130141-00-3 130141-01-4

RL: USES (Uses)

(thermal transition temps. of liq. cryst., fiber spinning in relation to)

RN 130141-00-3 HCAPLUS

CN 2-Dodecenedioic acid, (E)-, polymer with (Z)-4,4'-azoxybis[3-methylphenol] (9CI) (CA INDEX NAME)

CM 1

CRN 78992-83-3 CMF C14 H14 N2 O3

Double bond geometry as shown.

CM 2

CRN 6402-36-4 CMF C12 H20 O4

Double bond geometry as shown.

RN 130141-01-4 HCAPLUS

CN Nonanedioic acid, polymer with 4,4'-azoxybis[3-methylphenol] (9CI) (CA INDEX NAME)

CM 1

CRN 119176-66-8 CMF C14 H14 N2 O3

CM 2

CRN 123-99-9 CMF C9 H16 O4

 $HO_2C-(CH_2)_7-CO_2H$

L26 ANSWER 13 OF 18 HCAPLUS COPYRIGHT 2003 ACS

AN 1986:524362 HCAPLUS

DN 105:124362

TI Ferroelectric liquid crystal display cells

IN Crossland, William Alden; Davey, Anthony Bernard; Bone, Matthew Francis

PA ITT Industries, Inc., USA; Deutsche ITT Industries G.m.b.H.

SO Eur. Pat. Appl., 19 pp.

CODEN: EPXXDW

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
ΡI	EP 179592	A2	19860430	EP 1985-307174	19851008
	EP 179592	A3	19870916		
	R: AT, BE,	CH, DE	, FR, IT, LI,	LU, NL, SE	
	AU 8548516	A1	19860501	AU 1985-48516	19851011
	AU 573039	В2	19880526		
	JP 05004647	B4	19930120	JP 1985-237805	19851025
PRAI	GB 1984-26976		19841025		
7 D	7 bistable form		14		4 - 174 - 44

AB A bistable ferroelec. liq. crystal smectic I* or smectic F* display cell has a liq. crystal layer confined between opposed electrode plates. The inward facing surfaces are heated to provide planar alignment of the adjacent liq. crystal mols. in the same direction at each of the 2 surfaces. The plates serve to define a liq. crystal layer in the range 4-40 .mu. but which still permits bistable operation. Thus, a hermetically sealed envelope, for liq. crystal layer, was formed by securing together 2 glass sheets, of which the inward facing surfaces carried transparent In-Sn-oxide electrode layers covered with a polyimide layer. Both polyimide layers were rubbed in a single direction to promote planar alignment of the liq. crystal mols. in the direction of the rubbing and assembled parallel to each other. The cell was filled with the chiral ester C8H17C6H4-p-C6H4-p-C02C6H4-p-

CH2CH*(CH3)C2H5, also known as CE8. With the cell maintained at 69.8.degree., to keep the filling in I* phase, and mounted between crossed polarizers aligned with their polarization planes at 45.degree. to the rubbing direction, the pulse duration was .apprx.100 ms for 30 V pulses.

IC ICM G02F001-137

ICS G02F001-133; C09K019-20; C09K019-46

CC 74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
Section cross-reference(s): 75, 76

IT 54173-15-8 97139-97-4

RL: USES (Uses)

(ferroelec. liq. crystal layer contg., for display devices)

IT 70116-35-7

RL: USES (Uses)

(liq. crystals from, for display devices)

IT 54173-15-8 97139-97-4

RL: USES (Uses)

(ferroelec. liq. crystal layer contg., for display devices)

RN 54173-15-8 HCAPLUS

CN Benzenamine, N,N'-(1,4-phenylenedimethylidyne)bis[4-butyl-, [N(E),N'(E)]-(9CI) (CA INDEX NAME)

Double bond geometry as shown.

RN 97139-97-4 HCAPLUS

CN 1,4-Benzenediamine, N,N'-bis[[4-(heptyloxy)phenyl]methylene]-, [N(E),N'(E)]- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

PAGE 1-B

IT 70116-35-7

RL: USES (Uses)

(liq. crystals from, for display devices)

RN 70116-35-7 HCAPLUS

[1,1'-Biphenyl]-4-carboxylic acid, 4'-octyl-, 4-[(2S)-2-methylbutyl]phenyl CN ester (9CI) (CA INDEX NAME)

Absolute stereochemistry.

L26 ANSWER 14 OF 18 HCAPLUS COPYRIGHT 2003 ACS

AN1983:603667 HCAPLUS

DN 99:203667

Electrooptical display devices ΤI

Nippon Electric Co., Ltd., Japan PΑ

Jpn. Tokkyo Koho, 3 pp. SO CODEN: JAXXAD

DT Patent

LΑ Japanese

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE 19820717 JP 1973-113026 19731008 PΙ JP 57033565 B4 PRAI JP 1973-113026 19731008

Electrode plates for liq. crystal display devices are coated with .gtoreq.1 of 4-phenylurazole, isatin, 2,5-diketopiperazine, coumarin, 5-aminobarbituric acid, quinhydrone, 1,5-dihydroxyanthraquinone. The coatings promote homeotropic alignment of the liq.

crystal mols. in the display cell.

G02F001-133; C09K003-34 IC

CC 74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes) Section cross-reference(s): 75, 76

106-34-3 106-57-0 117-12-4 ΙT 91-56-5 91-64-5 118-78-5

15988-11-1 RL: USES (Uses)

(mol. orientation controlling films of, for liq. crystal display devices)

IT 15988-11-1

RL: USES (Uses)

(mol. orientation controlling films of, for liq. crystal display devices)

RN 15988-11-1 HCAPLUS

CN 1,2,4-Triazolidine-3,5-dione, 4-phenyl- (9CI) (CA INDEX NAME)

L26 ANSWER 15 OF 18 HCAPLUS COPYRIGHT 2003 ACS

AN 1980:13894 HCAPLUS

DN 92:13894

TI Induced smectic mesomorphism in mixtures of p-cyano-p'-pentylbiphenyl and p-pentylbenzoic acid

AU Yu, L. J.; Labes, M. M.

CS Dep. Chem., Temple Univ., Philadelphia, PA, 19122, USA

SO Molecular Crystals and Liquid Crystals (1979), 54(1-2), 1-8 CODEN: MCLCA5; ISSN: 0026-8941

DT Journal

LA English

AB The phase diagram of p-cyano-p'-pentylbiphenyl (CPB) and p-pentylbenzoic acid (5BA) shows a max. value of the nematic-isotropic transition temp. at apprx. equimolar concns. of CPB and the dimer of 5BA (D5BA). Two eutectic points occur at concns. of D5BA of 8 and 65 mol %, resp., and an induced smectic phase occurs at 15-65% D5BA. Dielec. properties and the spontaneous homeotropy of some of the mixts. can best be understood by considering the phases to consist primarily of D5BA and CPB, with a small amt. of free 5BA modulating the dielec. properties and promoting homeotropic alignment.

CC 75-4 (Crystallization and Crystal Structure)

Section cross-reference(s): 76

IT 26311-45-5

RL: PRP (Properties)

(induced sym. mesomorphism in mixt. of cyanopentylbiphenyl and dimeric)

IT 40817-08-1

RL: PRP (Properties)

(induced sym. mesomorphism in mixt. of pentylbenzoic acid and)

IT 26311-45-5

RL: PRP (Properties)

(induced sym. mesomorphism in mixt. of cyanopentylbiphenyl and dimeric)

RN 26311-45-5 HCAPLUS

CN Benzoic acid, 4-pentyl- (9CI) (CA INDEX NAME)

IT 40817-08-1

RL: PRP (Properties)

(induced sym. mesomorphism in mixt. of pentylbenzoic acid and)

RN 40817-08-1 HCAPLUS

CN [1,1'-Biphenyl]-4-carbonitrile, 4'-pentyl- (9CI) (CA INDEX NAME)

L26 ANSWER 16 OF 18 HCAPLUS COPYRIGHT 2003 ACS

AN 1977:585417 HCAPLUS

DN 87:185417

TI Role of coupling agents in surface modification of fillers

AU Plueddemann, E. P.; Stark, G. L.

CS Corp. Res., Dow Corning Corp., Midland, MI, USA

SO Modern Plastics (1977), 54(8), 76-8, 80 CODEN: MOPLAY; ISSN: 0026-8275

DT Journal

LA English

AB Silane coupling agents protect fillers against abrasion during polymer compounding, promote optimum alignment of polymer segments at interfaces, and overcome inhibitory catalytic effects fillers may have on polymer cure in addn. to promoting bonding at polymer-filler interfaces. The treated filler remains chem. inert during mixing, but combines with the polymer during molding to give optimum mech. strength and chem. resistance.

CC 36-6 (Plastics Manufacture and Processing)

IT 9003-53-6P 9011-14-7P

RL: SPN (Synthetic preparation); PREP (Preparation)

(prepn. of, filler effect on)

IT 9003-53-6P

RL: SPN (Synthetic preparation); PREP (Preparation)

(prepn. of, filler effect on)

RN 9003-53-6 HCAPLUS

CN Benzene, ethenyl-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 100-42-5 CMF C8 H8

 $H_2C = CH - Ph$

```
FEELY 09/899031
                  Page 53
```

ANSWER 17 OF 18 HCAPLUS COPYRIGHT 2003 ACS ΑN 1972:569882 HCAPLUS DN 77:169882 TIMolecular arrangement of nematic liquid crystals Uchida, Tatsuo; Watanabe, Hideo; Wada, Masanobu ΑU CS Dep. Electron. Eng., Tohoku Univ., Sendai, Japan SO Japanese Journal of Applied Physics (1972), 11(10), 1559-65 CODEN: JJAPA5; ISSN: 0021-4922 DTJournal English LAThe mol. arrangement of 12 types of nematic liq. crystals was obsd. on the AB surfaces of the supporting plates of glasses of soda, Pyrex, or quartz. These surfaces were treated by various methods. The impurity N-anisylidene-p-aminophenol or N-(p-ethoxybenzylidene)-p-aminophenol added to the nematic liq. crystals promotes the mol. alignment perpendicular to the glass plate. A mechanism of the mol. alignment of nematic liq. crystals on the glass surface is discussed in connection with the mol. structures of nematic liq. crystals based on the exptl. results. CC 70-4 (Crystallization and Crystal Structure) IT 1562-94-3 4792-83-0 10484-13-6 13036-19-6 24742-30-1 26227-73-6 29743-08-6 29743-09-7 30298-88-5 32185-12-9 32185-20-9 32185-21-0 RL: PRP (Properties) (lig. crystals of, mol. arrangement of nematic, glass substrate compn. effect on) IT 1562-94-3 4792-83-0 10484-13-6 13036-19-6 24742-30-1 26227-73-6 29743-08-6 29743-09-7 30298-88-5 32185-12-9 32185-20-9 32185-21-0 RL: PRP (Properties) (liq. crystals of, mol. arrangement of nematic, glass substrate compn. effect on) 1562-94-3 HCAPLUS RN CN Diazene, bis(4-methoxyphenyl)-, 1-oxide (9CI) (CA INDEX NAME)

4792-83-0 HCAPLUS RN CN Diazene, bis(4-ethoxyphenyl)-, 1-oxide (9CI) (CA INDEX NAME)

RN10484-13-6 HCAPLUS

CN Phenol, 4-[(E)-[(4-methoxyphenyl)methylene]amino]-, acetate (ester) (9CI) (CA INDEX NAME)

Double bond geometry as shown.

RN 13036-19-6 HCAPLUS

CN Benzonitrile, 4-[[(4-methoxyphenyl)methylene]amino]- (9CI) (CA INDEX NAME)

RN 24742-30-1 HCAPLUS

CN Benzonitrile, 4-[(E)-[(4-ethoxyphenyl)methylene]amino]- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

RN 26227-73-6 HCAPLUS

CN Benzenamine, 4-butyl-N-[(4-methoxyphenyl)methylene]- (9CI) (CA INDEX NAME)

$$N = CH$$

RN 29743-08-6 HCAPLUS

CN Benzenamine, 4-butyl-N-[(4-ethoxyphenyl)methylene]- (9CI) (CA INDEX NAME)

$$N = CH$$

RN 29743-09-7 HCAPLUS

CN Benzenamine, N-[(4-butoxyphenyl)methylene]-4-butyl- (9CI) (CA INDEX NAME)

RN 30298-88-5 HCAPLUS

CN Benzenamine, 4-butyl-N-[(4-propoxyphenyl)methylene]-, (E)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

RN 32185-12-9 HCAPLUS

CN Phenol, 4-[[(4-methoxyphenyl)methylene]amino]-, propanoate (ester) (9CI) (CA INDEX NAME)

RN 32185-20-9 HCAPLUS

CN Phenol, 4-[[(4-ethoxyphenyl)methylene]amino]-, acetate (ester) (9CI) (CA INDEX NAME)

RN 32185-21-0 HCAPLUS

CN Phenol, 4-[[(4-ethoxyphenyl)methylene]amino]-, propanoate (ester) (9CI) (CA INDEX NAME)

KATHLEEN FULLER EIC 1700/PARKER LAW 308-4290

L26 ANSWER 18 OF 18 HCAPLUS COPYRIGHT 2003 ACS

AN 1967:461176 HCAPLUS

DN 67:61176

TI Lipase from Candida paralipolytica. III. The activation of the enzyme systems with bile or calcium salts

AU Ota, Yasuhide; Yamada, Koichi

CS Univ. Tokyo, Tokyo, Japan

SO Agricultural and Biological Chemistry (1967), 31(7), 809-16 CODEN: ABCHA6; ISSN: 0002-1369

DT Journal

LA English

cf. CA 66: 26269g. The action of C. paralipolytica lipase in the in vitro AB hydrolysis of triglycerides was studied using reaction systems emulsified with 7 different emulsifiers. Systems emulsified with poly(vinyl alc.) required only Na taurocholate as essential activator. Systems emulsified with gelatin of methylcellulose required either Na taurocholate or Ca ions. In the presence of pectin or gum arabic emulsifiers, there was apparently no requirement for essential activators when olive oil was the substrate. Systems emulsified with Na cholate required the presence of NaCl. These variations in the emulsions were due to specific enzyme-emulsifier or substrate-emulsifier interaction. The systems also differed in their response to NaCl, which, at 0.2M concn., activated the gum arabic-emulsified system but completely inhibited that emulsified with lecithin and the poly(vinyl alc.) system activated with Na taurocholate. These results may be due to the difference in activation mechanisms by bile salts, Ca salts, and some macromol. emulsifiers. Various olive oil emulsions prepared with one of the above emulsifiers (except lecithin) were hydrolyzed at a similar rate at pH 8.2. The roles of bile salts and Ca salts during lipolysis by C. paralipolytica lipase are to promote the alignment of enzyme or substrate molecules and (or) the transcon formation of enzyme at the interface. references.

CC 3 (Enzymes)

IT 145-42-6 361-09-1 7440-70-2, biological studies
RL: BIOL (Biological study)

(lipase activation by)

IT 361-09-1

RL: BIOL (Biological study)
 (lipase activation by)

RN 361-09-1 HCAPLUS

CN Cholan-24-oic acid, 3,7,12-trihydroxy-, monosodium salt, (3.alpha.,5.beta.,7.alpha.,12.alpha.)- (9CI) (CA INDEX NAME)

Absolute stereochemistry.